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department at the University of Michigan from 1969 to 1972 and President of Hope College in  
Holland, Michigan, from 1972 to 1987. He was born in Grant, Michigan and turned 100 in February  
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Solution: Hg :  $L = 1 \text{ m}$ ;  $\rho = 13\,580 \text{ kg/m}^3$  from Table A.4 (or read Fig 2.7) The pressure difference  $\Delta P$  balances the column of height  $L$  so from Eq.2.2  $\Delta P = \rho g L = 13\,580 \text{ kg/m}^3 \times 9.80665 \text{ m/s}^2 \times 1.0 \text{ m} \times 10^{-3} \text{ kPa/Pa} = 133.2 \text{ kPa}$  Sonntag, Borgnakke and van Wylen 2.14 You dive 5 m down in the ocean.

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Solution:  $a = dV = \Delta V / \Delta t \Rightarrow \Delta t = \Delta V / a \Delta t = (50 - 15) \times 1609.34 \times 3.28084 / (3600 \times 15) = 3.42$  sec  
 $F = ma = 2500 \times 15 / 32.174 \text{ lbf} = 1165 \text{ lbf}$   
2.40E Two pound moles of diatomic oxygen gas are enclosed in a 20-lbm steel container. A force of 2000 lbf now accelerates this system.

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